## $Na_{4-x}Nb_7P_4O_{29}$ : A Phosphate Niobium Bronze Intergrowth of the Members m = 3 and m = 4 of the MPTB's Series $Na_x(NbO_3)_{2m}(PO_2)_4$

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A new niobium phosphate bronze  $Na_{4-x}Nb_7P_4O_{29}$  ( $0 \le x \le 1$ ) has been isolated. It crystallizes in the space group C2/c with a = 32.850(4), b = 5.3484(4), and c = 13.252(2) Å and  $\beta = 97.79(1)^\circ$ . The structure was solved by X-ray diffraction on a twinned crystal with an (100) twinning plane. This oxide is an intergrowth of the m = 3 and m = 4 members of the series of the monophosphate tungsten bronzes with pentagonal tunnels (MPTB<sub>P</sub>), corresponding to the formulation  $Na_x(NbO_3)_{2m}(PO_2)_4$ . The ReO<sub>3</sub> slabs alternate parallel to (100) and are linked by isolated PO<sub>4</sub> tetrahedra delimiting pentagonal tunnels running along **b**. Relationships with other niobium phosphate bronzes are also discussed. @ 1992 Academic Press, Inc.

### Introduction

The systematic study of tungsten phosphates, started in the 1980s, allowed a great number of phosphate tungsten bronzes with various structures to be isolated (1-2). These phases were classified into four series: the monophosphate and the diphosphate tungsten bronzes with hexagonal tunnels noted (MPTB<sub>H</sub> and DPTB<sub>H</sub>, respectively) and those with pentagonal tunnels  $(MPTB_p \text{ and } DPTB_p)$ . More recently, the synthesis of the bronzoid phosphate  $\beta$ -NbPO<sub>5</sub> (3), isotypic of WPO<sub>5</sub> (4), second member of the  $MPTB_{P}$  family  $(WO_3)_{2m}(PO_2)_4$ , has allowed a similar behavior to be expected for niobium phosphate bronzes. However, in the other members of this family, the valency of tungsten is always greater than 5, so that the synthesis of other

niobium phosphates, with formulation  $(NbO_3)_{2m}(PO_2)_4$ , is not possible for m > 2. Nevertheless the fact that the MPTB<sub>n</sub>'s exhibit empty pentagonal tunnels suggests that cations of intermediate size such as sodium could be inserted into the structure. allowing the synthesis of bronzes with the general formula  $Na_x(NbO_3)_{2m}(PO_2)_4$  to be considered. The recent synthesis of the monophosphate niobium bronzes  $Na_{2+r}$  $Nb_6P_4O_{26}$  (5) and  $Na_4Nb_8P_4O_{32}$  (6), corresponding to the members m = 3 and m = 4of this family, respectively, confirms this viewpoint and opens the road to the research of a complete series of monophosphate niobium bronzes with the MPTB<sub>p</sub> structure. We report here on the crystal structure of the bronze  $Na_{4-r}Nb_7P_4O_{29}$ , which corresponds to m = 3.5, intergrowth of the members m = 3 and m = 4.

#### Synthesis and Characterization

The bronzes  $Na_{4-x}Nb_7P_4O_{29}$  were prepared in two steps. First the adequate mixture of  $Na_2CO_3$ ,  $Nb_2O_5$ , and  $H(NH_4)_2PO_4$  was ground in an agate mortar and heated up to 673 K. Then, the appropriate amount of niobium was added to the intermediate composition and the mixture was sealed in an evacuated silica ampoule and heated at 1343 K during 3 days, allowing dark blue crystals to be grown and a pure phase to be prepared.

The limit phase, x = 1, which corresponds to the bronzoid Na<sub>3</sub>Nb<sub>7</sub>P<sub>4</sub>O<sub>24</sub>, was prepared by a similar method but entirely in air. The first step of the reaction was performed at the same temperature of 673 K. Then the mixture was reground and heated in a platinum crucible at 1373 K in air for 1 day. Under these conditions, the bronzoid is absolutely pure. Its X-ray powder pattern was indexed in a monoclinic cell

in agreement with the parameters obtained from the single crystal study (Table I).

#### **Structure Determination**

All the crystals obtained from the preparation were found to be twinned. A dark blue crystal of dimensions  $0.103 \times 0.077 \times 0.025$  mm was selected for the structure determination.

A careful study of the  $(010)^*$  reciprocal levels, registered with a Weissenberg camera, led to the following results.

-This twinned crystal was likely constituted of two components as suggested by electron microscopy investigations on several samples. The relative disposition of the reflections issued from the two twin components is illustrated in Fig. 1, taking as an example the *h2l* level.

—For each component, the 2/m Laue

| hkl     | $d_{\rm obs}$ | $d_{\rm calc}$ | Ι   | h k l    | $d_{\rm obs}$ | $d_{\rm obs}$ | Ι  |
|---------|---------------|----------------|-----|----------|---------------|---------------|----|
| 40 0    | 8.155         | 8.137          | 28  | 2 2 0    | 2.638         | 2.638         | 36 |
| 40-2    | 5.480         | 5.486          | 12  | 120-2    |               | 2.636         |    |
| 402     | 4.807         | 4.801          | 33  | 314      |               | 2.636         |    |
| 310     |               | 4.797          |     | 11 1 0   | 2.589         | 2.589         | 79 |
| 60-2    | 4.494         | 4.492          | 66  | 100 - 4  | 2.485         | 2.486         | 13 |
| 11 - 2  | 4.160         | 4.158          | 69  | 11 1 1   |               | 2.485         |    |
| 112     | 4.070         | 4.069          | 85  | 12 0 4   | 1.964         | 1.964         | 24 |
| 80 0    |               | 4.068          |     | 11 1 - 5 |               | 1.964         |    |
| 31-2    | 3.990         | 3.989          | 100 | 13 1 3   | 1.916         | 1.919         | 26 |
| 602     | 3.928         | 3.928          | 71  | 82 - 4   |               | 1.915         |    |
| 312     | 3.767         | 3.767          | 43  | 624      | 1.883         | 1.883         | 20 |
| 80 - 2  | 3.687         | 3.690          | 40  | 22 - 5   |               | 1.882         |    |
| 51-2    | 3.642         | 3.641          | 60  | 13 0     | 1.778         | 1.780         | 43 |
| 512     | 3.371         | 3.369          | 52  | 11 - 7   |               | 1.780         |    |
| 802     | 3.264         | 3.265          | 32  | 11 1 -6  |               | 1.779         |    |
| 40-4    | 3.204         | 3.198          | 37  | 716      |               | 1.777         |    |
| 100 - 2 | 3.087         | 3.088          | 18  | 531      | 1.694         | 1.697         | 32 |
| 712     | 2.970         | 2.971          | 15  | 102 - 5  |               | 1.696         |    |
| 114     | 2.761         | 2.760          | 37  | 42 - 6   |               | 1.694         |    |
| 80 - 4  | 2.746         | 2.743          | 27  | 02 - 6   |               | 1.693         |    |
| 020     | 2.676         | 2.674          | 64  |          |               |               |    |

TABLE I Na:Nb:P.O...-Powder Pattern



FIG. 1. The reciprocal plane  $(h2l)^*$ . ( $\bigstar$ ) First component, ( $\bigcirc$ ) second component.

symmetry is involved. Besides main reflections, l = 2n, weak reflections, l = 2n + 1, were observed; they are indicative of a superstructure of the second order along the **c** axis. Two space groups, Cc and C2/c, are consistent with conditions h + k = 2n for hkl and l = 2n for h0l.

—The correspondence between the two crystalline components can be described either from a (100) symmetry plane or from a [001] twofold axis.

—The reflections l = 3n were found to be quite superimposed (see Fig. 1), owing to the values of the a,c, and  $\beta$  parameters of the monoclinic unit cell; the relative deviation  $(3c^*\cos\beta^* - a^*)/a^*$  is about 0.01. These reflections exhibit a symmetry very close to the upper orthorhombic symmetry.

—The  $\alpha$  proportion of the smallest component was estimated from the intensity measurement of 50 strong pairs of nonoverlapped reflections; the dispersion of the results is small,  $\alpha = 0.393(8)$ , bearing out the twinning hypothesis.

The data of one of the components were collected on a CAD-4 Enraf-Nonius diffractometer, the parameters of the monoclinic unit cell (Table II) refined by least squares from the  $\theta$  measurements of 25 nonoverlapped reflections ( $18^{\circ} \le \theta \le 25^{\circ}$ ); the nearly perfect superimposition of the reflections l = 3n was confirmed from an  $\omega - \theta$  profile study. The reflections were corrected for Lorentz and polarization effects. No absorption corrections were performed.

According to the features of the Patterson series, the space group is C2/c. The structure was solved using in a first step the 1666 nonoverlapped reflections. The positions of the niobium atoms were determined from the Patterson peaks and the other atoms located from subsequent Fourier difference series. The refinements, carried out with anisotropic thermal parameters for Nb and P

| TA | BL | Æ | Η |
|----|----|---|---|
|    |    |   |   |

SUMMARY OF CRYSTAL DATA, INTENSITY MEASUREMENTS, AND STRUCTURE REFINEMENT PARAMETERS FOR Na<sub>3.04</sub>Nb<sub>7</sub>P<sub>4</sub>O<sub>79</sub>

| 1. Crystal data                 |                                    |  |  |  |
|---------------------------------|------------------------------------|--|--|--|
| Space group                     | C2/c                               |  |  |  |
| Cell dimensions                 | a = 32.850(4)  Å                   |  |  |  |
|                                 | b = 5.3484(4)  Å                   |  |  |  |
|                                 | c = 13.252(2)  Å                   |  |  |  |
|                                 | $\beta = 97.79(1)^{\circ}$         |  |  |  |
| Volume                          | $v = 2307(1) Å^3$                  |  |  |  |
| Ζ                               | 4                                  |  |  |  |
| 2. Intensity measurements       |                                    |  |  |  |
| $\lambda$ (MoK $\alpha$ )       | 0.71073 Å                          |  |  |  |
| Scan mode                       | $\omega - \theta$                  |  |  |  |
| Scan width (°)                  | $1 + 0.35 \tan \theta$             |  |  |  |
| Slit aperture (mm)              | $1 + \tan \theta$                  |  |  |  |
| $Max \theta$ (°)                | 45°                                |  |  |  |
| Standard reflections            | 3 measured every 3600 s (no decay) |  |  |  |
| Reflections with $I > 3 \sigma$ | 2655                               |  |  |  |
| 3. Structure solution and refi  | nement                             |  |  |  |
| Parameters refined              | 202                                |  |  |  |
| Agreement factors               | $R = 0.052$ $R_w = 0.069$          |  |  |  |
| Weighting scheme                | w = 1                              |  |  |  |
| $\Delta/\sigma$ max             | 0.007                              |  |  |  |
| $\Delta/\sigma$ max             | 0.007                              |  |  |  |

atoms and isotropic ones for the other atoms, led to the reliability factors R = 0.044 and  $R_w = 0.067$ .

In a second step, the superimposed reflections (989) were considered. The intensities of such reflections have generally to be corrected for the twinning effect from the relations (7):

$$J_1 = I_1 + \frac{\alpha}{1 - 2\alpha} (I_1 - I_2);$$
  
$$J_2 = I_2 - \frac{\alpha}{1 - 2\alpha} (I_1 - I_2).$$

 $I_1$  and  $I_2$  are the observed intensities of the superimposed reflection pairs of the two components of the twin, the indices of which are, respectively, hkl and (-h + 2l/3), k, lwith l = 3 n;  $J_1$  and  $J_2$  are the corrected intensities, i.e., the intensities which would be observed from a single crystal of the same total volume. Since the intensities  $I_1$  and  $I_2$  are nearly equal, no correction was applied to this group of reflections. The refinements were carried out with all the reflections using the CLINUS program (8) and introducing different scale factors for the superimposed and nonoverlapped reflections. The final reliability factors are R = 0.052 and  $R_w = 0.069$ . The atomic parameters are reported in Table III. The refinement of the multiplicity of Na sites shows that Na(1) was fully occupied and Na(2) only partially occupied, leading to the formula Na<sub>3.04</sub>Nb<sub>7</sub>P<sub>4</sub>O<sub>29</sub>.

# Description of the Structure and Discussion

This study shows that a regular intergrowth of the members m = 3 and m = 4 of the series MPTB<sub>P</sub>'s has been obtained

| Atom      | x          | У         | Z          | $B(Å^2)$ |
|-----------|------------|-----------|------------|----------|
| <br>Nb(1) | 0.000      | 0.000     | 0.000      | 0.73(2)  |
| Nb(2)     | 0.31831(3) | 0.5011(4) | 0.02287(7) | 0.43(1)  |
| Nb(3)     | 0.22724(3) | 0.4785(3) | 0.32190(8) | 0.48(1)  |
| Nb(4)     | 0.04595(2) | 0.4973(4) | 0.17911(7) | 0.50(1)  |
| P(1)      | 0.14822(8) | 0.487(1)  | 0.1205(2)  | 0.47(4)  |
| P(2)      | 0.42169(7) | 0.503(1)  | 0.1525(2)  | 0.39(3)  |
| Na(1)     | 0.1318(2)  | 0.507(2)  | 0.3634(4)  | 1.28(9)  |
| Na(2)     | -0.0976(4) | 0.009(4)  | -0.0995(9) | 2.0(2)   |
| O(1)      | -0.0468(3) | -0.025(3) | 0.0780(8)  | 1.6(2)   |
| O(2)      | 0.0270(4)  | -0.286(2) | 0.0731(9)  | 1.2(2)   |
| O(3)      | 0.0299(4)  | 0.207(2)  | 0.1076(8)  | 1.2(2)   |
| O(4)      | 0.3791(3)  | 0.459(2)  | 0.0956(8)  | 1.1(2)   |
| O(5)      | 0.3066(3)  | 0.730(2)  | 0.1235(8)  | 0.8(2)   |
| O(6)      | 0.3024(3)  | 0.227(2)  | 0.0847(7)  | 0.6(1)   |
| O(7)      | 0.2710(3)  | 0.564(2)  | -0.0695(7) | 0.7(1)   |
| O(8)      | 0.3401(3)  | 0.278(2)  | -0.0883(7) | 0.8(1)   |
| O(9)      | 0.3482(3)  | 0.821(2)  | -0.0378(7) | 1.0(2)   |
| O(10)     | 0.1737(3)  | 0.562(2)  | 0.2216(8)  | 1.0(2)   |
| O(11)     | 0.2478(3)  | 0.291(2)  | 0.2264(7)  | 0.9(1)   |
| O(12)     | 0.000      | 0.515(5)  | 0.250      | 1.6(3)   |
| O(13)     | 0.1029(3)  | 0.480(3)  | 0.1375(7)  | 1.2(2)   |
| O(14)     | 0.0722(4)  | 0.798(2)  | 0.2645(9)  | 1.4(2)   |
| O(15)     | 0.0755(4)  | 0.264(2)  | 0.3037(8)  | 1.1(2)   |

TABLE III Positional Parameters and Their Estimated Standard Deviations

*Note.* Anisotropically refined atoms are given in the form of the isotropic equivalent displacement parameter, defined as  $(4/3) * [a^2 * B(1,1) + b^2 * B(2,2) + c^2 * B(3,3) + ab(\cos \gamma) * B(1,2) + ac(\cos \beta) * B(1,3) + bc(\cos \alpha) * B(2,3)]$ .

in the form of a pure phase for the first time. A similar phenomenon was observed in the case of tungsten phosphate bronzes

 $(WO_3)_{2m}(PO_2)_4$ , but only for an integral m value (m = 5), i.e.,  $P_4W_{10}O_{38}$  (10) intergrowth of the members m = 4 and m =



FIG. 2. Projection of the  $Na_{4-x}Nb_7P_4O_{29}$  structure along **b.** ( $\bullet$ ) Na(1) and ( $\blacksquare$ ) Na(2).

| Nb(1)                 | O(1)                  | <b>O</b> (1 <sup>i</sup> ) | O(2)                 | <b>O</b> (2 <sup>i</sup> ) | O(3)     | O(3 <sup>i</sup> )    |
|-----------------------|-----------------------|----------------------------|----------------------|----------------------------|----------|-----------------------|
| O(1)                  | 1.97(1)               | 3.94(2)                    | 2.80(2)              | 2.75(2)                    | 2.79(2)  | 2.77(2)               |
| O(1 <sup>i</sup> )    | 180.0(7)              | 1.97(1)                    | 2.75(2)              | 2.80(2)                    | 2.77(2)  | 2.79(2)               |
| O(2)                  | 91.2(6)               | 88.8(6)                    | 1.96(1)              | 3.92(2)                    | 2.68(2)  | 2.86(2)               |
| O(2 <sup>i</sup> )    | 88.8(6)               | 91.2(6)                    | 180.0(7)             | 1.96(1)                    | 2.86(2)  | 2.68(2)               |
| O(3)                  | 90.4(5)               | 89.6(5)                    | 86.2(5)              | 93.8(5)                    | 1.96(1)  | 3.92(2)               |
| O(3 <sup>i</sup> )    | 89.6(5)               | 90.4(5)                    | 93.8(5)              | 86.2(5)                    | 180.0(7) | 1.96(1)               |
| Nb(2)                 | O(4)                  | O(5)                       | O(6)                 | O(7)                       | O(8)     | O(9)                  |
| O(4)                  | 2.108(9)              | 2.85(1)                    | 2.79(1)              | 3.95(1)                    | 2.77(1)  | 2.72(1)               |
| O(5)                  | 91.0(4)               | 1.89(1)                    | 2.74(1)              | 2.81(1)                    | 3.97(1)  | 2.73(2)               |
| O(6)                  | 91.1(4)               | 96.2(5)                    | 1.79(1)              | 2.81(1)                    | 2.76(1)  | 3.96(1)               |
| O(7)                  | 165.5(4)              | 96.6(4)                    | 100.3(4)             | 1.873(9)                   | 2.77(1)  | 2.87(1)               |
| O(8)                  | 82.4(4)               | 170.8(4)                   | 90.3(4)              | 88.5(4)                    | 2.09(1)  | 2.98(2)               |
| O(9)                  | 78.9(4)               | 84.1(4)                    | 169.9(4)             | 89.7(4)                    | 88.4(4)  | 2.18(1)               |
| Nb(3)                 | O(5 <sup>ii</sup> )   | O(6 <sup>iii</sup> )       | O(7 <sup>iv</sup> )  | O(10)                      | O(11)    | O(11 <sup>iii</sup> ) |
| O(5 <sup>ii</sup> )   | 1.93(1)               | 2.70(1)                    | 2.78(1)              | 2.72(1)                    | 2.87(2)  | 3.91(1)               |
| O(6 <sup>iii</sup> )  | 83.1(4)               | 2.14(1)                    | 2.85(1)              | 2.72(1)                    | 3.94(1)  | 2.79(2)               |
| O(7 <sup>iv</sup> )   | 92.9(4)               | 89.6(4)                    | 1.905(9)             | 3.99(1)                    | 2.82(1)  | 2.82(1)               |
| O(10)                 | 84.8(4)               | 79.9(4)                    | 169.5(4)             | 2.104(9)                   | 2.83(1)  | 2.85(1)               |
| O(11)                 | 99.7(5)               | 171.4(4)                   | 98.4(4)              | 92.1(4)                    | 1.82(1)  | 2.74(1)               |
| O(11 <sup>iii</sup> ) | 166.6(5)              | 84.6(4)                    | 92.3(4)              | 87.9(4)                    | 91.8(5)  | 2.00(1)               |
| Nb(4)                 | O(2 <sup>v</sup> )    | O(3)                       | O(12)                | O(13)                      | O(14)    | O(15)                 |
| O(2 <sup>v</sup> )    | 1.86(1)               | 2.75(2)                    | 2.82(2)              | 2.82(2)                    | 2.80(2)  | 4.04(2)               |
| O(3)                  | 95.2(5)               | 1.86(1)                    | 2.78(2)              | 2.79(2)                    | 3.93(2)  | 2.83(1)               |
| O(12)                 | 97.8(6)               | 96.0(8)                    | 1.885(2)             | 3.88(2)                    | 2.80(2)  | 2.82(2)               |
| O(13)                 | 92.9(5)               | 91.8(6)                    | 166.1(3)             | 2.02(1)                    | 2.68(2)  | 2.74(2)               |
| O(14)                 | 90.1(5)               | 171.7(5)                   | 89.5(7)              | 81.6(5)                    | 2.09(1)  | 2.90(2)               |
| O(15)                 | 173.2(5)              | 88.5(5)                    | 87.5(6)              | 81.2(5)                    | 85.6(4)  | 2.19(1)               |
| P(1)                  | O(8 <sup>vi</sup> )   |                            | O(9 <sup>vii</sup> ) | O(10)                      |          | O(13)                 |
| O(8 <sup>vi</sup> )   | 1                     | .55(1)                     | 2.54(2)              | 2.                         | 54(1)    | 2.48(2)               |
| O(9 <sup>vii</sup> )  | 112                   | .1(6)                      | 1.52(1)              | 2.                         | 52(1)    | 2.46(2)               |
| O(10)                 | 110.8(6)              |                            | 111.6(6)             | 1.53(1)                    |          | 2.48(1)               |
| O(13)                 | 107.3(7)              |                            | 107.1(7)             | 107.6(6)                   |          | 1.54(1)               |
| P(2)                  | O(1 <sup>viii</sup> ) |                            | O(4)                 | O(14 <sup>ii</sup> )       |          | O(15 <sup>iii</sup> ) |
| O(1 <sup>viii</sup> ) | 1.53(1)               |                            | 2.48(1)              | 2.53(2)                    |          | 2.48(2)               |
| O(4)                  | 108                   | .8(6)                      | 1.52(1)              | 2.                         | 43(1)    | 2.48(1)               |
| O(14 <sup>ii</sup> )  | 110                   | .9(8)                      | 105.4(6)             | 1.                         | 54(1)    | 2.54(2)               |
| O(15 <sup>iii</sup> ) | 109                   | .0(8)                      | 109.9(6)             | 112.                       | 7(6)     | 1.51(1)               |

TABLE IV Distances (Å) and Angles (°) in the NbO, and PO, Polyhedra and Main Na-O Distances

*Note.* The Nb–O or P–O distances are on the diagonal. Above it are the O(i) . . . O(j) distances and below it are the O(i)–Nb–P–O(j) or O(i)–O(j) angles. Na(1)–O(4<sup>ii</sup>) = 3.01(1) Å; Na(2)–O(1) = 2.70(2) Å; Na(1)–O(4<sup>ii</sup>) = 2.51(1) Å; Na(2)–O(2<sup>i</sup>) = 2.74(2) Å; Na(1)–O(5<sup>ii</sup>) = 2.50(1) Å; Na(2)–O(3<sup>i</sup>) = 2.52(2) Å; Na(1)–O(6<sup>iii</sup>) = 2.47(1) Å; Na(2)–O(4<sup>ix</sup>) = 2.81(2) Å; Na(1)–O(9<sup>ii</sup>) = 2.52(1) Å; Na(2)–O(8<sup>ix</sup>) = 2.41(2) Å; Na(1)–O(10) = 2.49(1) Å; Na(2)–O(9<sup>ix</sup>) = 2.65(2) Å; Na(1)–O(13) = 3.02(1) Å; Na(2)–O(13<sup>i</sup>) = 2.66(3) Å; Na(1)–O(14) = 2.70(1) Å; Na(2)–O(13<sup>x</sup>) = 2.78(3) Å; Na(1)–O(15) = 2.31(1) Å; Na(2)–O(14<sup>x</sup>) = 2.65(3) Å.

Symmetry code. i: -x, -y, -z; vi:  $\frac{1}{2} - x, \frac{1}{2} - y, -z;$  ii:  $\frac{1}{2} - x, -\frac{1}{2} + y, \frac{1}{2} - z;$  vii:  $\frac{1}{2} - x, \frac{3}{2} - y, -z;$  iii:  $\frac{1}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z;$  viii:  $\frac{1}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z;$  viii:  $\frac{1}{2} + x, \frac{1}{2} + y, z;$  iv:  $x, 1 - y, \frac{1}{2} + z;$  ix:  $-\frac{1}{2} + x, -\frac{1}{2} + y, z;$  v: x, 1 + y, z; x: -x, 1 - y, -z.

6. In contrast, intergrowths corresponding to nonintegral *m* values were only observed by HREM in microcrystals of tungsten phosphates but could not be isolated in a quantitative manner. Thus the framework of the nobium bronze phosphate  $Na_{4-x}$  $Nb_7P_4O_{29}$  (Fig. 2) is built up from the same kind of ReO<sub>3</sub>-type slabs as those previously observed in  $Na_{2+x}Nb_6P_4O_{26}$  (5) and  $Na_4$   $Nb_8P_4O_{32}$  (6). These slabs are, respectively, three and four corner-sharing octahedra wide oriented approximately along the direction [104] for the former and [103] for the latter. The junction between these slabs is ensured by slices of isolated PO<sub>4</sub> tetrahedra, giving rise to pentagonal tunnels running along **b**.

The geometry of NbO<sub>6</sub> octahedra is quite



FIG. 3. The two twin components with the glide plane parallel to (100) situated between (a) two blocks of three octahedra and (b) two blocks of four octahedra.

similar to that of the other members of this series,  $Na_{2+r}Nb_6P_4O_{26}$  (4) and  $Na_4Nb_8P_4O_{32}$ (5). In both structures, the octahedra located at the border of the ReO<sub>3</sub> slabs are more distorted than the octahedra disposed inside the same block (Table IV). For example, in this structure the Nb-O distances range from 1.96 to 1.97 Å for Nb(1) and from 1.82 to 2.14 Å for Nb(3); moreover the disparity between the Nb-O distances of the corresponding octahedra in the same block is more pronounced, ranging from 1.86 to 2.19 Å for Nb(4) and from 1.79 to 2.18 Å for Nb(2). The distortion of each NbO<sub>6</sub> octahedron increases with the number of PO<sub>4</sub> tetrahedra sharing its corners. The PO4 tetrahedra are nearly regular as in other phosphate bronzes. These results confirm again that the PO<sub>4</sub> tetrahedra, more rigid, impose a rather strong distortion to the octahedral framework in order to realize such structures. The comparison of the geometry of NbO<sub>6</sub> octahedra with that of WO<sub>6</sub> octahedra in the various members of the series  $(WO_3)_{2m}(PO_2)_4$  (9–11) is in agreement with this viewpoint. One indeed observes for the latter the same kind of distortion of the  $WO_6$ octahedra at the border and inside the ReO<sub>3</sub>type slabs. However, the variation of the amplitude of the distortion is smaller for the  $WO_6$  octahedra than for the NbO<sub>6</sub> octahedra, owing to the larger size of niobium compared to tungsten. This difference of size seems to be at the orgin of the inability of niobium to form diphosphate niobium bronzes similar to those of tungsten.

The sodium atoms have the same environment as that of the previous members of this series. They are located approximately at the center of the hexagonal windows situated at the boundary between two adjacent tunnels and formed of three octahedra and three tetrahedra. The two sodium sites are ninefold coordinated to oxygen atoms. The great tendancy of this phase to be twinned is explained by the existence of a local pseudo-symmetry described by a glide plane *n* parallel to (100) located at  $x = \frac{1}{8}$ , with a translation part of c/4 along c.

Examples of such twinning planes are shown in Fig. 3; such models show that the twinning takes place at the level of the rows of pentagonal tunnels, i.e., at the level of the slices of  $PO_4$  tetrahedra which border the ReO<sub>3</sub>-type slabs.

#### **Concluding Remarks**

This study confirms the existence of series of monophosphate niobium bronzes with pentagonal tunnels, with the general formula  $Na_x$  (NbO<sub>3</sub>)<sub>2m</sub>(PO<sub>2</sub>)<sub>4</sub>, belonging to the MPTB<sub>P</sub>'s structural family. The electron transport properties of these materials will be investigated.

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